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**AIMS** - To investigate bed sediment-water interactions during summer (a critical period for P effects on ecosystems) and compare P sorption properties of sediments and soils.

**SAMPLING** - Agricultural soils, river water and bed sediments were sampled from tributaries and main stem sites of the River Dee (NE Scotland) during: (i) May-04 (storm flow), (ii) Aug-04 (extended baseflow period), and (iii) May-05 (flow event recession).

### MODELLING P SORPTION:

(i) Equilibrium P concentration,  $EPC_0$  (solution concentration at which no net sorption occurs) derived from batch isotherms ( $0-20 \mu\text{M PO}_4\text{(P)}$  in  $0.01\text{M CaCl}_2$ , at  $5^\circ\text{C}$ ) according to  $n_a = K_d c - n_i$ , where  $K_d$  is the distribution coefficient,  $n_a$  and  $n_i$  the adsorbed and original (FeO paper desorbable P) sediment/soil P content,  $c$  is the equilibrium concentration of SRP, then  $EPC_0 = c$  when  $\Delta n_a = 0$ .

(ii) The molar ratio,  $Z = P_{ox} / 0.5(\text{Fe}_{ox} + \text{Al}_{ox})$  acid ammonium oxalate extractable Fe, Al, P.

(iii)  $P = \text{sediment } EPC_0 - \text{river [SRP]}$  (+ve values = P release).

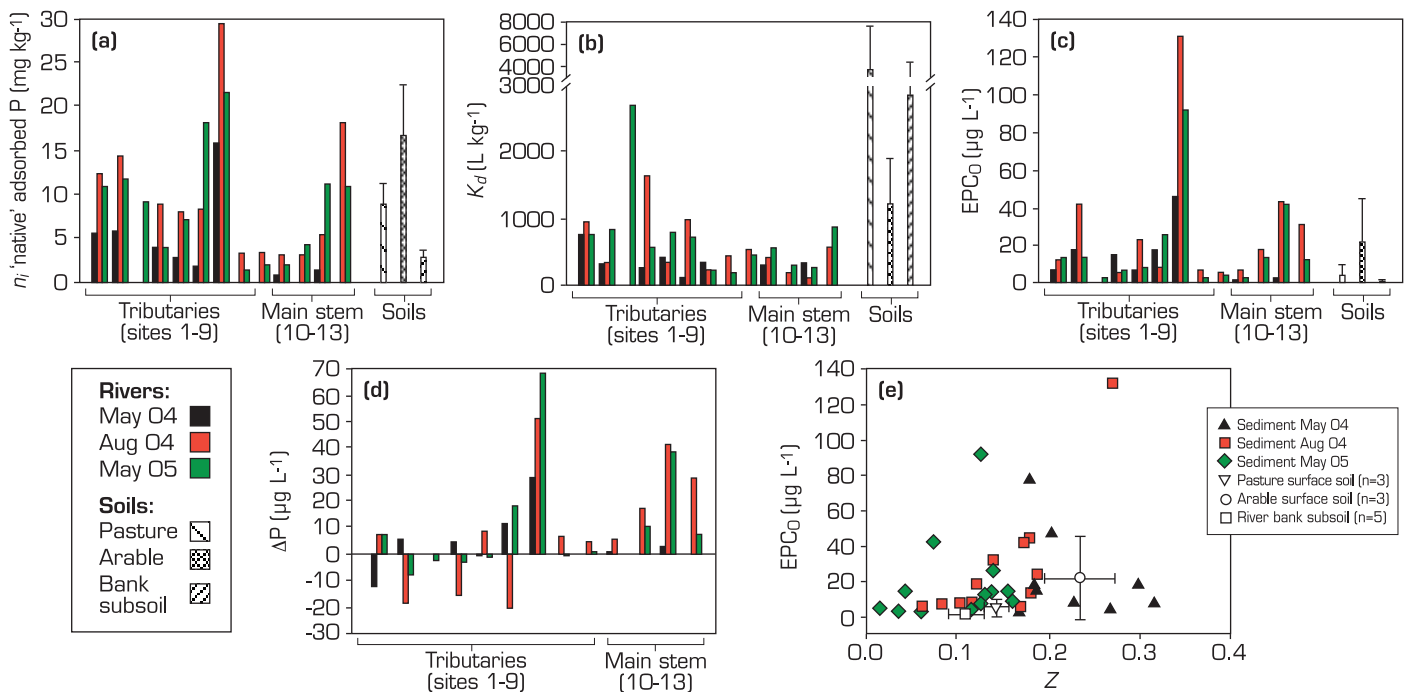
### HOW DO LAND USE PRESSURES AFFECT SEDIMENT P STATUS?

Native sediment P ( $n_i$ ), desorption potential ( $EPC_0$ ) and P saturation ( $Z$ ) increased from: low intensity land use sites (tributaries 8, 9, main stem 10, 11) < downstream main stem sites (12, 13) < agricultural tributaries (1-6) and << than in the presence of a point source (tributary 7, STW population 1200).

Table 1. Catchment, temporal mean river data and spatial mean soil properties

Site	Catchment		River water			Sediment / soil	
	Area (km <sup>2</sup> )	Pasture + arable % area	pH	SRP ( $\mu\text{g l}^{-1}$ )	pH <sub>water</sub>	Total Ca (g kg <sup>-1</sup> )	Org C (g kg <sup>-1</sup> )
1	4	46	7.49	10	6.48	16	15
2	31	67	7.51	32	6.40	20	20
3	35	32	7.16	5	6.21	11	28
4	37	63	7.53	14	6.42	22	13
5	51	60	7.49	11	6.54	31	5
6	58	63	7.46	15	6.57	27	5
7	71	56	7.48	41	6.35	11	13
8	94	2	6.50	2	5.93	11	7
9	212	10	6.70	4	5.63	6	5
10	1005	3	6.84	2	6.03	13	8
11	1348	9	6.97	2	6.23	9	6
12	1775	13	7.05	3	5.97	18	6
13	2005	19	7.13	3	6.04	11	42
Pasture surface soil						5.03	41
Arable surface soil						5.88	34
Stream bank subsoil						5.76	28

Fig. 1. Sorption properties (a)  $n_i$ , (b)  $K_d$ , (c)  $EPC_0$  for river bed sediments and soils, (d) P values for sediments, and (e)  $EPC_0$  against  $Z$  (site numbers as in Table 1).



### ARE SEDIMENTS SOURCES OR SINKS OF P TO WATERS?

Sediments were mostly potential P sources to waters (Fig. 1d). In 5, 10 and 18 cases  $\Delta P$  values ( $\mu\text{g P l}^{-1}$ ) were respectively categorized as <5 (sink), -5 to 5 (in equilibrium) and >5 (source). Lower reaches of the main stem had accumulated sediment P likely to desorb ( $EPC_0$  values  $\gg$  river water SRP concentrations). However, during baseflow (Aug-04) bed sediments became potentially strong sinks for P at the most intensively agricultural sites (sites 2, 4, 6).

### HOW DO RIVER SEDIMENT P SORPTION PROPERTIES COMPARE TO THOSE OF SOURCE SOILS?

Smaller  $K_d$  values for sediments suggested weaker SRP sorption to sediments than soils. Smaller sediment  $EPC_0$  and larger  $Z$  values in May-04 showed that sediments became more similar to arable topsoil at high flow. Conversely, increased sediment  $EPC_0$  and  $n_i$  baseflow suggested in-stream accumulation of P. Fig. 1e shows how the sediment P release potential varies with surface P saturation. In Aug-04 a sharp increase in  $EPC_0$  for  $Z > 0.12$  is similar to the 'change point concept' applied to soils (three outlying data points relate to larger Ca contents and potentially stronger P sorption related to clay content).

### HOW CAN MODELLING APPROACHES BE IMPROVED?

Simplistic modelling methods highlight spatio-temporal variability in P interactions between sediments and waters related to diffuse and point source pollution pressures and in-stream processing. However, simulations may be refined by incorporation of the effects of biotic processes, redox and changing water chemistry in P modelling approaches.